

IN VITRO ANTIOXIDANT ACTIVITY OF ETHANOLIC EXTRACT OF BARK OF ALSTONIA SCHOLARIS

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ABSTRACT

Alstonia scholaris belonging to family Apocynaceae is commonly known as Devil's tree or Saptaparni. It is an evergreen tropical tree native to Indian sub-continent and South East Asia. It contains various phytoconstituents like alkaloids, triterpenoids, flavonoids, steroids and phenolic acids which have shown promising therapeutic potential. The present research was subjected to evaluate the in vitro antioxidant activity of ethanolic extract of bark of *Alstonia scholaris* using 2,2-diphenyl-1-picrylhydrazyl (DPPH) radical scavenging assay. The extract showed significant antioxidant activity when compared to ascorbic acid. The results of this research work are promising thus indicating the utilisation of the bark of *Alstonia scholaris* as a significant source of natural antioxidant.

Keywords: *Alstonia scholaris*, antioxidant activity, DPPH free radical scavenging activity

I. INTRODUCTION

Due to their versatile applications, plant derived substances are gaining great interest and there has been an increased focus on plant research all over the world. *Alstonia scholaris* belonging to family Apocynaceae is popularly known as “Devil’s tree” or “Saptaparni” and is attaining attention of researcher’s for its pharmacological activities. The plant has been used in Ayurvedic, Unani, Homeopathy and Folklore system of medicine to treat various types of disorders.^[1,2] It is widely distributed in dried forests of India, Western Himalayas, and Western Ghats as well as in the Southern India^[3]. It has been reported to possess antimalarial^[4-5], antimicrobial^[6-7], free radical scavenging and antioxidant^[8-11], anti-diabetic^[12-14], analgesic and anti-inflammatory^[15-16], anticancer and cytotoxicity^[17-22], radioprotective^[23-25], CNS activity^[26-27],

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immunostimulating^[28-29], antifertility^[30], antidiarrheal^[31-32], bronchodilatory^[33], anti-tussive and anti-asthmatic^[34] activities.

Phytoconstituents have been reported in different parts of the plant such as bark;^[35-36] leaves;^[37-38] roots;^[39] flowers^[40] and fruits^[41]. *Alstonia scholaris* is known to be a rich source of alkaloids (about 180 alkaloids)^[42-43] like Echitamine^[35,44], Scholarine^[45], Picrinine^[46], corialstonine and corialstonidine^[47]. Isolation of a new secoiridoid glucoside alstonoside, together with two known isoflavone apioglucosides, formononetin 7-0-β-D-apiofuranosyl(1-6)-β-D-glucopyranoside and biochanin α-7-0-β-D apiofuranosyl(1-6)-β-D-glucopyranoside has also been reported^[48]. Among the other constituents, Isookanin-7-o-alpha-l-rhamnopyranoside, a new flavanone glycoside^[48] and Alstonoside, a secoiridoid glucoside^[49] has been recorded. Iridoids, coumarins, flavonoids, leucoanthocyanins, reducing sugars, simple phenolics, steroids, saponins and tannins were also found in the plant^[50]. Total phenolics, flavonoid and tannin contents were found to be significantly (P<0.05) higher in ethanolic extracts of *Astonia scholaris*^[51-52].

The aim of this paper is to analyze the antioxidant activity of *Alstonia scholaris* bark. Antioxidants are molecules that are capable of stabilizing, inhibiting, and deactivating free radicals before reacting^[53-54]. They are divided into endogenous antioxidants like superoxide dismutase (SOD), glutathione peroxidase (GPX), and catalase (CAT) that play an important role in maintaining good cellular function and health; and exogenous antioxidants. However, because of oxidative stress, body needs exogenous antioxidants to compensate it^[54-55]. These are further divided into natural antioxidants and synthetic antioxidants. As the use of synthetic antioxidants (BHA, BHT, propyl gallate) are not recommended in pharmacology due to safety concerns^[56] thus many natural antioxidants from fruits and plants need to be developed. Compounds that act as natural antioxidants are vitamins, phenolics, flavonoids, carotenoids and thiols^[57-58].

II. MATERIALS AND METHODS

Chemicals and plant material

1,1-Diphenyl-2-picrylhydrazyl (DPPH) was obtained from Sigma Chemicals Co., USA. Ascorbic acid and all other chemicals and solvents used were of analytical grade.

Collection of plant part

The mature bark from woody trunk portion of *Alstonia scholaris* was procured from local area of Ghaziabad, India and the species of the plant was authenticated by the Botanist of Raw Material Herbarium and Museum (RHMD), Delhi (taxonomic reference number: RHMD-3653-54)

Preparation of extract

The collected bark was cleaned, washed, air dried and coarsely powdered. The weight of the dry powder was taken. The coarsely powdered bark (100 gm) was taken in a 500 ml soxhlet apparatus and extracted with 85% ethanol for around 24 hours. After extraction, the ethanol extract was dried free of solvent in a rotary evaporator at low temperature (40°C). This dried extract was tested for antioxidant activity.

Antioxidant assay

DPPH free radical scavenging assay

The antioxidant activity was measured using the standard DPPH method. The capacity of the extracts to scavenge the stable free radical DPPH was monitored according to the procedure described by Amarowicz et al^[59]. DPPH being a stable free radical accepts an electron or hydrogen atom to become 1,1-diphenyl-2-picrylhydrazine molecule. The purple colour of free DPPH changes to yellow upon reaction with hydrogen donors because of reaction between hydrogen donors and DPPH free radical that leads to its reduction to corresponding hydrazine. Various concentrations (2, 4, 8, 16, 32, 64 and 128 µg/ml) of the extract were added to 4 ml of a 0.004 % ethanol solution of DPPH. The mixture was shaken and left for 25-30 mins at room temperature in the dark. The reduction of DPPH radical was determined by reading the decrease in absorbance at 517 nm using an ultraviolet-visible spectrophotometer. All determinations were performed in triplicate. The antioxidant activity was calculated as the percent inhibition caused by the hydrogen donor activity of each sample according to the following formula:

$$(\%) \text{ Inhibition} = (1 - \text{Absorbance of the sample} / \text{Absorbance of the Blank}) \times 100$$

III. RESULT AND DISCUSSION

From table 1, it was observed that the ethanolic bark extract of *Alstonia scholaris* showed the highest antioxidant activity of 80.3% as compared to the standard ascorbic acid, 73.6% by DPPH free radical scavenging method (figure 1). This study showed that the ethanolic extract has proton donating ability and could serve as a free radical inhibitor or scavenger and act as a natural antioxidant.

Table 1: DPPH free radical scavenging activity of *Alstonia scholaris* bark extract

Sam ple No.	Concentratio n (µg/mL)	%Inhibition	
		Ethanolic extract <i>A. scholaris</i>	Ascorbic acid
1.	2	42.6	40.4
2.	4	46.3	50.2
3.	8	50.8	58.3
4.	16	57.4	62.5
5.	32	66.7	65.8
6.	64	72.4	68.2
7.	128	80.3	73.6

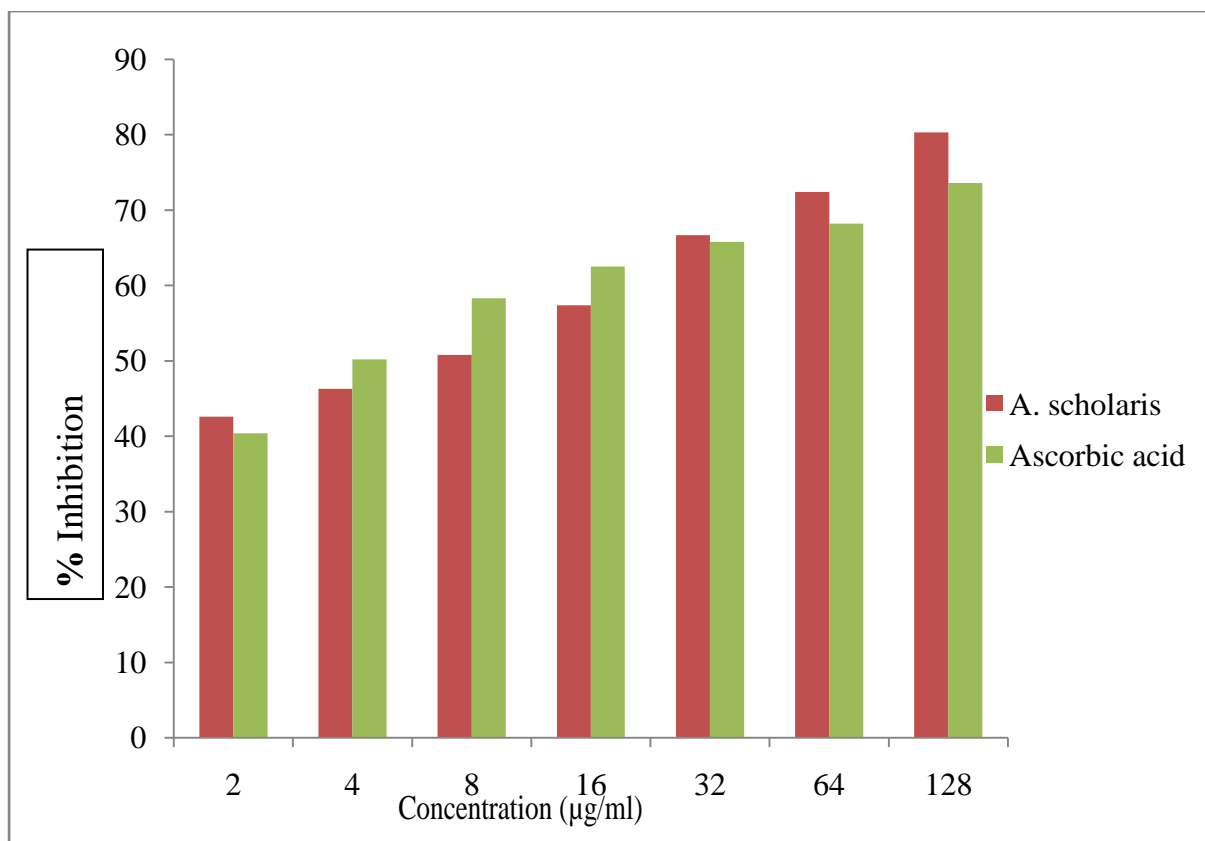


Figure 1: Percentage of free radicals scavenged by ethanolic extract from bark of *A. scholaris* using Ascorbic acid as standard (DPPH free radical scavenging method)

IV. CONCLUSION

In vitro antioxidant activity of the bark extract of *Alstonia scholaris* was performed. The results of DPPH assay shows that the extract has antioxidant activity when compared to the standard ascorbic acid indicating that this extract can be a significant source of natural antioxidants, which might be helpful in preventing the diseases involving free radicals and can be further used for the benefit of mankind.

CONFLICT OF INTEREST

No conflict of interest lies between Authors.

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